1.6, 7.3), 3.10 (br s, 1), 3.30–3.50 (br s, 1), 3.51 (dd, 1, J = 1.6, 9.1), 4.01 (s, 1). ¹³C NMR (75 MHz, C_6D_6): δ 9.7, 18.9, 19.5, 26.3, 30.7, 36.0, 45.7, 75.4, 82.3, 220.0.

(1R,2R,4S)-1,4-Dihydroxy-2,5,5-trimethyl-1-phenyl-3hexanone (20d). ¹H NMR (300 MHz): δ 0.94 (s, 9), 1.09 (d, 3, J = 7.2, 3.14–3.25 (br m, 2), 3.25–3.38 (br s, 1), 3.93 (s, 1), 5.11 (d, 1, J = 3.8), 7.25–7.40 (m, 5). ¹³C NMR (75 MHz): δ 11.2, 26.2, 36.0, 50.8, 72.5, 82.4, 126.1, 127.6, 128.2, 141.2, 218.6.

(3S,5S,6S)-3,6-Dihydroxy-2,2,5,7-tetramethyl-4-octanone (21b). ¹H NMR (400 MHz): δ 0.88 (d, 3, J = 6.8), 0.93 (d, 3, J= 6.8), 0.95 (s, 9), 0.97 (d, 3, J = 6.9), 1.82 (double septet, 1, J= 6.9, 2.1, 2.93-2.94 (m, 1), 3.29-3.36 (m, 1), 3.49-3.53 (m, 1), 3.67 (d, 1, J = 2.3), 4.49 (br s, 1). ¹³C NMR (100 MHz): δ 12.46, 14.47, 19.78, 26.38, 29.95, 35.89, 47.26, 80.34, 85.58, 218.1.

(1S,2R,4S)-1,4-Dihydroxy-2,5,5-trimethyl-1-phenyl-3hexanone (21d). ¹H NMR (250 MHz): δ 0.76 (d, 3, J = 6.9), 0.96 (s, 9), 3.38 (d, 1, J = 2.8), 3.42-3.52 (m, 1), 3.72 (d, 1, J = 2.8) 2.6), 4.53 (d, 1, J = 2.7), 4.55 (d, 1, J = 2.7), 7.35–7.36 (m, 5). ¹³C NMR (75 MHz): δ 13.37, 26.15, 35.27, 48.31, 78.62, 84.70, 126.86, 128.36, 128.54, 141.75, 217.70.

(3S,5S,6S)-3,6-Dihydroxy-2,2,5,7-tetramethyl-4-octanone (22b). ¹H NMR (400 MHz): δ 0.88 (d, 3, J = 6.7), 0.96 (d, 3, J= 6.9, 0.99 (s, 9), 1.01 (d, 3, J = 6.8), 1.77 (dq, 1, J = 3.5, 6.9), 2.28-2.30 (m, 1), 3.04 (d pent, 1, J = 1.7, 6.8), 3.41 (d, 1, J = 7.1),3.43-3.47 (m, 1), 4.04 (dd, 1, J = 1.2, 7.1). ¹³C NMR (100 MHz): δ 12.50, 14.53, 19.77, 26.39, 29.99, 35.91, 47.19, 80.41, 85.61

(1R,2S,4S)-1,4-Dihydroxy-2,5,5-trimethyl-1-phenyl-3hexanone (22d). ¹H NMR (300 MHz): δ 0.87 (d, 3, J = 6.7), 0.99 (s, 9), 2.62 (br s, 1), 3.17 (dq, 1, J = 8.8, 6.7), 3.19 (d, 1, J= 2.0), 4.06 (s, 1), 4.59 (d, 1, J = 8.8), 7.30–7.35 (m, 5). ¹³C NMR (75 MHz): § 12.78, 26.38, 35.89, 51.40, 79.27, 85.80, 126.57, 128.32, 128.57, 142.09, 218.2.

(-)-(2S,3R)-3-Hydroxy-2,4-dimethylpentanoic Acid (23b). $[\alpha]_{\rm D}$: -9.5° (c = 0.4, H₂Cl22). Th ¹H NMR spectrum of this material was identical to that reported.²⁶ ¹³C NMR (50 MHz): δ 14.59, 16.09, 19.75, 30.67, 42.65, 78.04, 180.96.

(-)-(2S,3S)-3-Hydroxy-2-methyl-3-phenylpropanoic Acid (23d). $[\alpha]_D$: -29.3° (c = 0.8, CHCl₃). [lit. $[\alpha]_D$ -29.5° (c = 2.03, CHCl₃)].²⁷ ¹H NMR (400 MHz): δ 1.16 (d, 3, J = 7.2), 2.85 (dq, 1, J = 3.9, 7.2, 5.19 (d, 1, J = 3.9), 7.27–7.37 (m, 5). ¹³C NMR

(26) Montgomery, S. H.; Pirrung, M. C.; Heathcock, C. H. Organic Synthyses; Wiley: New York, 1990, Collect. Vol. VII, p 190.

(100 MHz): δ 10.27, 46.16, 73.37, 125.93, 127.66, 128.35, 141.01, 180.85.

(+)-(2R,3S)-3-Hydroxy-2,4-dimethylpentanoic Acid (24b). $[\alpha]_D$ +9.1° (c = 2.2, CHCl₃). The ¹H and ¹³C NMR spectra of this material were identical with those obtained for its enantiomer, 23b.

(+)-(2R,3R)-3-Hydroxy-2-methyl-3-phenylpropanoic Acid (24d). $[\alpha]_{\rm D}$: +28.5° (c = 1.2, CHCl₃). [lit. $[\alpha]_{\rm D}$ +29.5° (c = 1.27, CHCl₂)].²⁶ The ¹H and ¹³C NMR spectra were identical with those of its enantiomer, 23d.

(-)-(2R,3R)-3-Hydroxy-2,4-dimethylpentanoic Acid (25b). $[\alpha]_{D}$: -14.3° (c = 1.0, CHCl₃). The ¹H NMR spectrum of this material was identical with that reported.¹⁸ ¹³C NMR (50 MHz): δ 9.69, 18.72, 19.02, 30.62, 41.77, 76.93, 181.27.

(-)-(2R,3S)-3-Hydroxy-2-methyl-3-phenylpropanoic Acid (25d). $[\alpha]_{D}$: -17.5° (c = 2.3, CHCl₃). ¹H NMR (400 MHz): δ 1.00 (d, 3, J = 7.2), 2.84 (dq, 1, J = 7.2, 9.0), 4.75 (d, 1, J = 9.0), 5.70–6.10 (br s, 1), 7.29–7.39 (m, 5). ¹³C NMR (100 MHz): δ 14.40, 47.20, 76.60, 126.84, 128.40, 128.80, 141.16, 180.72. Compound 25d was identified by comparison of its ¹H NMR spectrum with that reported.28

(+)-(2S,3S)-3-Hydroxy-2,4-dimethylpentanoic Acid (26b). $[\alpha]_{D}$: +14.1° (c = 1.1, CHCl₃). The ¹H and ¹³C NMR spectra of this material were identical with those obtained for the enantiomer. 25b.

(+)-(2S,3R)-3-Hydroxy-2-methyl-3-phenylpropanoic Acid (26d). $[\alpha]_{D}$: +17.8° (c = 2.0, CHCl₃). The ¹H and ¹³C NMR spectra were identical with those of the enantiomer, 25d.

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Supplementary Material Available: Figures 2 and 3 (ORTEP drawings of keto diols 19 and 21) and ¹H and ¹³C NMR spectra of keto diols 19b, 19d, 20b, 20d, 21b, 21d, 22b, and 22d (17 pages). Ordering information is given on any current masthead page.

Stereocontrolled Preparation of *cis*- and *trans*-2,6-Dialkylpiperidines via 1-Acyldihydropyridine Intermediates. Synthesis of (\pm) -Solenopsin A and (±)-Dihydropinidine

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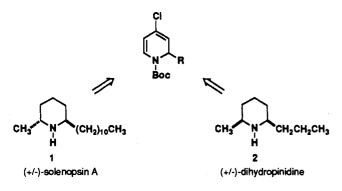
The stereoselective reduction of 1-(tert-butoxycarbonyl)-4-chloro-2,6-dialkyl-1,2-dihydropyridines 6 and 22 was studied. Reduction of 6 with Et₃SiH/TFA gave the cis-2,6-dialkyl-1,2,5,6-tetrahydropyridine 7 as the major product. The stereoselectivity was reversed by reducing 6 with NaBH₃CN/TFA, which gave predominantly the trans-2,6-dialkyltetrahydropyridine 10. Catalytic hydrogenation of 7 and 10 gave the corresponding N-Boc-cis(or trans)-2,6-dialkylpiperidines. Regioselective hydrogenation of 6 gave the 1,2,3,4-tetrahydropyridine 18, which on treatment with NaBH₃CN/TFA provided a 90:10 mixture of trans- and cis-piperidines 15 and 16. More vigorous hydrogenation of 6 afforded the cis-piperidine 15 with 96% stereoselectivity. Similar stereoselective reductions of dihydropyridine 22 were carried out. Stereoselective reductions of dihydropyridines 6 and 22 were utilized in the synthesis of (±)-solenopsin A and (±)-dihydropinidine from 4-chloropyridine in six and five steps, respectively.

Alkaloids containing a 2,6-disubstituted piperidine ring are abundant in nature and many exhibit significant biological activity.² Numerous cis-2,6-disubstituted piperidines can be stereoselectively prepared by simple reduction

⁽²⁷⁾ Heathcock, C. H.; White, C. T.; Morrison, J. J.; VanDerveer, D.

J. Org. Chem. 1981, 46, 1296. (28) Evans, D. A.; Nelson, J. V.; Vogel, E.; Taber, T. R. J. Am. Chem. Soc. 1981, 103, 3099.

of substituted pyridine derivatives or by various intramolecular cyclizations, although frequently mixtures of cis and trans isomers result.² Stereoselective syntheses of trans-2,6-disubstituted piperidines are generally more difficult and include the hydride reduction of cyclic imines,³ alkene nitrone cycloadditions,⁴ alkylation and reduction of cyanopiperidines⁵ or bicyclic carbamates,⁶ the imine-epoxide rearrangement,⁷ and the alkylation of α lithiated piperidine derivatives.⁸ As part of a program directed at developing the utility of 1-acyldihydropyridines as synthetic intermediates,⁹ we investigated a strategy for the selective preparation of cis- and trans-2,6-disubstituted piperidines from a common 1-acyl-2-alkyl-1,2-dihydropyridine intermediate.¹⁰ We chose the well-characterized alkaloids (\pm) -solenopsin A $(1)^2$ and (\pm) -dihydropinidine $(2)^2$ as targets for this study. In this paper we report the development of methodology that allows the stereoselective preparation of cis- or trans-2,6-disubstituted piperidines from readily prepared 1-acyl-1,2-dihydropyridine intermediates.



Results and Discussion

Our approach to the *trans*-2,6-dialkylpiperidine (\pm) solenopsin A utilized 4-chloropyridine as starting material (Scheme I). Recent work from our laboratories has shown that the addition of Grignard reagents to the 1-phenoxycarbonyl salt of 4-chloropyridine gives 1-acyl-1,2-dihydropyridines in good to excellent yield.¹¹ A mixture of 4-chloropyridine (3) and undecylmagnesium bromide in tetrahydrofuran (THF) at -78 °C was treated with phenyl chloroformate to give the 1,2-dihydropyridine 4. Crude 4 was converted to the N-Boc derivative 5 with potassium *tert*-butoxide in THF in 86% overall yield from 3. A methyl group was introduced at C6, using directed-lithiation methodology.^{11a} Treatment of 5 with n-butyllithium

(1) Address correspondence to this author at Department of Chemistry, North Carolina State University, Raleigh, NC 27695-8204.

(2) Fodor, G. B.; Colasanti, B. In Alkaloids: Chemical and Biological Perspectives; Pelletier, S. W., Ed.; Wiley: New York, 1985; Vol. 3, pp 1-90. Baliah, V.; Jeyaraman, R.; Chandrasekaran, L. Chem. Rev. 1983, 83, 379.

(3) Yamamoto, H.; Maruoka, K.; Matsumura, Y. Tetrahedron Lett. 1982, 23, 1929; see also ref 10.

(4) Carruthers, W.; Williams, M. Chem. Commun. 1986, 1287. Tufariello, J.; Puglis, J. Tetrahedron Lett. 1986, 27, 1489. Gallagher, T.; Lathbury, D. Tetrahedron Lett. 1985, 26, 6249.

(5) Husson, H.-P.; Bonin, M.; Romero, J.; Grierson, D. J. Org. Chem. 1984, 49, 2392. Takahashi, K.; Kurita, H.; Ogura, K.; Iida, H. J. Org. Chem. 1985, 50, 4368.

(6) Ciufolini, M.; Hermann, C.; Whitmire, K.; Byrne, N. J. Am. Chem. Soc. 1989, 111, 3473.

(7) Wasserman, H. H.; Rodriques, K.; Kucharczyk, R. Tetrahedron Lett. 1989, 30, 6077.

(9) Comins, D. L.; Myoung, Y.-C. J. Org. Chem. 1990, 55, 292. Comins, D. L.; Goehring, R. R.; Joseph, S. P.; O'Connor, S. J. Org. Chem. 1990, 55, 2574 and references therein.

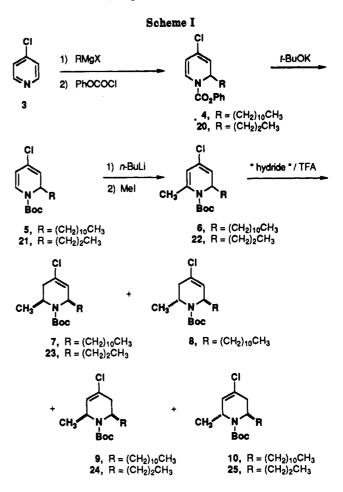


Table I. Stereoselective Reduction of Dihydropyridine 6

entry ^a	reducing agent ^b	conditions	yield,° %	ratio ^d 7:8:9:10
a	Et ₃ SiH	10.0 TFA, -42 °C CH ₂ Cl ₂ , 4 h	79	75:0:10:15
Ъ	n-Pr ₃ SiH	10.0 TFA, -42 °C CH ₂ Cl ₂ , 4 h	95°	78:0:9:13
с	NaBH ₃ CN	10.0 TFA, -42 °C CH ₂ Cl ₂ , 4 h	55	12:4:7:77

^aReactions were generally performed on a 0.2-1.0-mmol scale. ^bSix equiv. of reducing agent were used. ^cYield of isolated products 7, 8, 9, and 10 obtained from MPLC. ^dRatio determined by GC. ^eYield of crude reaction product.

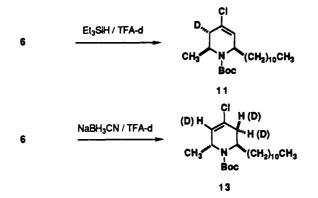
and methyl iodide gave an 83% yield of dihydropyridine 6. At this point the synthetic strategy called for stereoselective reduction of the C5–C6 double bond in 6. Reaction of 6 with trialkylsilane/trifluoroacetic acid (TFA)^{11a} or NaBH₃CN/TFA gave a crude product mixture that was analyzed (GC) for compounds 7–10. As is shown in Table I, the major product from the triethylsilane reduction was the *cis*-tetrahydropyridine 7. Interestingly, minor products (9 and 10) derived from double-bond migration were present as well. Use of the slightly larger tri-*n*-propylsilane effected a minor increase in the cis product 7. *Remarkably*, *the analogous reduction using sodium cyanoborohydride*

⁽¹⁰⁾ Yamaguchi and co-workers have carried out a similar strategy preparing both *cis*- and *trans*-2-methyl-6-octylpiperidine selectively from a common 2,6-disubstituted 1,2-dihydropyridine. Nakazono, Y.; Yamaguchi, R.; Kawanisi, M. *Chem. Lett.* 1984, 1129. Yamaguchi, R.; Nakazono, Y.; Matsuki, T.; Hata, E.; Kawansi, M. *Bull. Chem. Soc. Jpn.* 1987, 60, 215.

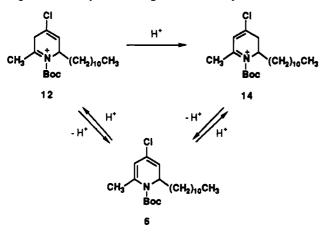
^{(11) (}a) Comins, D. L.; Weglarz, M. A.; O'Connor, S. Tetrahedron Lett. 1988, 29, 1751. (b) Comins, D. L.; Mantlo, N. B. J. Org. Chem. 1985, 50, 4410.

as the reducing agent gave a reversal in stereoselectivity with double-bond migration, producing the trans-tetrahydropyridine 10 as the major product.

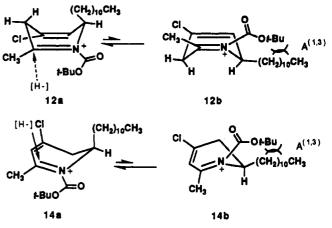
The reduction of 6 with triethylsilane and deuterated TFA (TFA-d) gave C5-deuterated tetrahydropyridine 11 as the major product, indicating that N-acyliminium ion 12 is the reactive intermediate in the formation of 7 from 6 and triethylsilane/TFA. The analogous reaction using



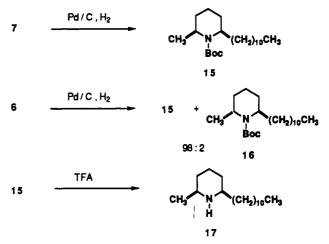
NaBH₃CN and deuterated TFA gave 13 as the major product, which had partial deuterium incorporation at C3 and C5. This result suggests that N-acyliminium ion 14 is the reactive species in this reduction and ion 12 is not readily reduced by NaBH₃CN. N-Acyliminium ion 12 is undoubtedly present in situ, but in the presence of NaB-H₃CN it may revert back to dihydropyridine 6 at a faster rate than reduction, or a rapid acid-catalyzed double-bond migration¹² may occur to give 14 directly from 12.

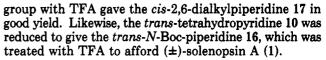


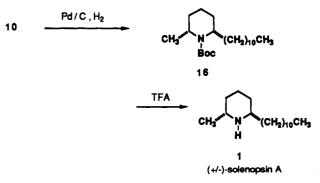
The stereochemical outcome of these reductions can be explained in the following manner. N-Acyliminium ion 12 must exist in a boat conformation. Of the two possible boats 12a and 12b, 12a is of much lower energy due to the additional $A^{(1,3)}$ strain¹³ present in conformation 12b. "Bottom side" attack of hydride on acyliminium ion 12a leads to the observed *cis*-tetrahydropyridine 7. The Nacyliminium ion 14 exists as a flat chair with the 2-substituent in the axial orientation due to $A^{(1,3)}$ strain. Stereoelectronically preferred¹⁴ axial attack of hydride at C6 of 14a leads to the observed *trans*-tetrahydropyridine 10 as the major product.¹⁵



The tetrahydropyridine 7 was reduced by catalytic hydrogenation to give carbamate 15. The cis product 15 could be prepared more conveniently by catalytic reduction of dihydropyridine 6, which gave a 98:2 mixture of cis and trans isomers 15 and $16.^{16}$ Removal of the Boc protecting







An alternative synthesis of 1 was devised that proved to be more convenient and gave a higher overall yield of

⁽¹²⁾ A similar acid-catalyzed rearrangement of an analogous N-alkyliminium ion has been proposed. Fry, E. M. J. Org. Chem. 1964, 29, 1647.

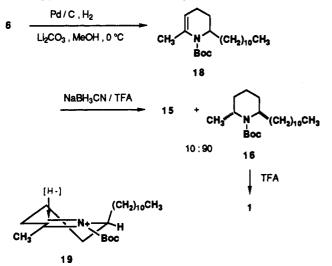
 ⁽¹³⁾ Hoffman, R. W. Chem. Rev. 1989, 89, 1841-1860. Johnson, F. Chem. Rev. 1968, 68, 375-413. Brown, J. D.; Foley, M. A.; Comins, D. L. J. Am. Chem. Soc. 1988, 110, 7445 and references therein.

⁽¹⁴⁾ Deslongchamps, P. Stereoelectronic Effects in Organic Chemistry; Pergamon: New York, 1983; Chapter 6.

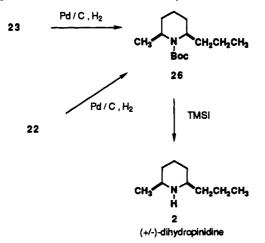
⁽¹⁵⁾ For related stereocontrolled additions of nucleophiles to 1-acyl-piperidinium ions, see: Comins, D. L.; Foley, M. A. Tetrahedron Lett.
1988, 29, 6711. Shono, T.; Matsumura, Y.; Onomura, O.; Sato, M. J. Org. Chem. 1988, 53, 4118. Hanson, G. J.; Russell, M. A. Tetrahedron Lett.
1989, 30, 5751. Natsume, M.; Sekine, Y.; Ogawa, M.; Soyagimi, H.; Kitagawa, Y. Tetrahedron Lett. 1979, 3473. Hiemstra, H.; Speckamp, W. N. In The Alkaloids; Brossi, A., Ed.; Academic Press, Inc.: San Diego, 1988; Vol. 32, pp 271-339, and references therein.

⁽¹⁶⁾ Yamaguchi and co-workers have reduced an N-(methoxycarbonyl)-6-methyl-2-alkynyl-1,2-dihydropyridine to a cis-N-(methoxycarbonyl)-2,6-dialkylpiperidine in two steps using catalytic hydrogenation. See ref 10.

the target compound. Dihydropyridine 6 was partially reduced to tetrahydropyridine 18 in 76% yield. When 18 was subjected to NaBH₃CN/TFA reduction, a 90:10 mixture of trans to cis products (16 and 15) resulted. The trans stereochemistry results from stereoelectronically preferred¹⁴ axial attack of hydride on the chair iminium ion 19, which contains the undecyl group in the axial orientation due to $A^{(1,3)}$ strain.^{13,15} With use of this route, (±)-solenopsin A (1) was prepared in six steps from 4chloropyridine with an overall yield of 35%.



The cis-2,6-dialkylpiperidine (\pm) -dihydropinidine (2) was prepared in a similar manner to the preparation of 17. A mixture of 4-chloropyridine (3) and *n*-propylmagnesium chloride in THF at -78 °C was treated with phenyl chloroformate to give the 1,2-dihydropyridine 20. Crude 20 was converted to the N-Boc derivative 21 with potassium *tert*-butoxide in 93% overall yield from 3. A methyl group was introduced at C6 in 84% yield via directed lithiation to give 22. Reduction of 22 with tri-n-propylsilane/TFA gave tetrahydropyridines 23, 24, and 25 in a ratio of 80/ 8/12.17 The cis-tetrahydropyridine 23 was isolated from this mixture by chromatography in 72% yield. Catalytic hydrogenation of 23 gave carbamate 26, which on treatment with TMSI afforded (\pm) -dihydropinidine (2) in 58% yield for the two steps. Alternatively, dihydropyridine 22 could be hydrogenated with 96% stereoselectivity to give cis-piperidine derivative 26 in 69% yield.



In summary, the combination of directed lithiation/ alkylation of 2-alkyl-N-Boc-4-chloro-1,2-dihydropyridines and subsequent reduction with tri-*n*-propylsilane/TFA, NaBH₃CN/TFA, or H₂, Pd/C is effective for the preparation of *cis*- or *trans*-2,6-dialkylpiperidine derivatives. The versatility of this strategy was demonstrated by the synthesis of (\pm) -solenopsin A¹⁸ and (\pm) -dihydropinidine¹⁹ from 4-chloropyridine in six and five steps, respectively.

Experimental Section

Reactions were performed in oven-dried glassware under an atmosphere of dry argon and were magnetically stirred. Tetrahydrofuran (THF) was dried by distillation from sodium benzophenone ketyl under nitrogen immediately prior to use. Other solvents were dried over 3-Å or 4-Å molecular sieves prior to use. Radial preparative-layer chromatography (radial PLC) was performed on a Chromatotron (Harrison Associates, Palo Alto, CA) using glass plates coated with 1, 2, or 4 mm thicknesses of Kieselgel 60 PF₂₅₄ containing gypsum.

4-Chloro-1-(phenoxycarbonyl)-2-n-undecyl-1,2-dihydro**pyridine** (4). To a stirred mixture of magnesium turnings (292) mg, 12.0 mmol) in 20 mL of anhydrous diethyl ether was added 1-bromoundecane (2.68 mL, 12.0 mmol). After an initial selfsustained reflux period of 10 min, the solution was refluxed an additional 45 min and then allowed to cool to room temperature. The newly formed Grignard reagent was transferred dropwise via a double-tipped needle into a stirred solution of 4-chloropyridine (0.946 mL, 10.0 mmol) in 80 mL of THF at -78 °C. After 20 min, phenyl chloroformate (1.26 mL, 10.0 mmol) was added dropwise. Stirring was continued for 30 min at -78 °C. The cooling bath was removed and the reaction mixture was allowed to stir for 30 min longer while being slowly warmed to room temperature. Aqueous 20% NH₄Cl (30 mL) and ether (40 mL) were added, the layers were separated, and the aqueous phase was extracted with ether $(2 \times 25 \text{ mL})$. The combined organic extracts were washed successively with 25-mL portions of saturated aqueous CuSO4, water, saturated aqueous NaHCO₃, and brine. The organic phase was dried over MgSO₄, filtered through Celite, and concentrated in vacuo to give 4.053 g (quantitative) of the crude 1,2-dihydropyridine 4 as a clear yellow oil (This crude material was used directly in the next step to make the N-Boc derivative.): ¹H NMR (300 MHz, CDCl₃) δ 7.45-7.10 (m, 5 H), 6.95-6.85 (pair of d due to rotamers, 1 H, J = 7.8 Hz), 5.68 (d, 1 H, J = 6.3 Hz), 5.40–5.28 (pair of dd due to rotamers, 1 H, J = 7.8 Hz and J = 2.0 Hz), 5.00-4.85 (pair of m due to rotamers, 1 H), 1.80-1.40 (m, 2 H), 1.25 (br s, 18 H), 0.88 (t, 3 H, J = 6.8 Hz); FT-IR (neat) 2954, 2923, 2852, 1735, 1635, 1592, 1495, 1471, 1332, 1202, 1050 cm⁻¹.

1-(tert-Butoxycarbonyl)-4-chloro-2-n-undecyl-1,2-dihydropyridine (5). To a stirred solution of crude dihydropyridine 4 (4.053 g, 10.39 mmol) in 120 mL of THF at -42 °C was added dropwise over 15 min a solution of potassium tert-butoxide (4.60 g, 41.0 mmol) in 50 mL of THF. The resulting orange solution was stirred for 1 h at -42 °C (complete reaction verified by TLC). The cooling bath was removed and the reaction mixture was allowed to stir for 20 min while being slowly warmed to room temperature. Water (30 mL) and ether (60 mL) were added, and the aqueous phase was extracted with ether $(2 \times 10 \text{ mL})$. The combined organic extracts were washed with cold 1 N NaOH (2 \times 25 mL), water (2 \times 25 mL), and brine. The organic phase was dried over MgSO₄, filtered through Celite, and concentrated in vacuo to give the crude product. Purification by radial PLC (silica gel, 5% EtOAc/hexanes) gave 3.20 g (86%) of 5 as a colorless oil: ¹H NMR (300 MHz, $CDCl_3$) δ 6.90-6.60 (pair of br d due to rotamers, 1 H, J = 7.5 Hz), 5.55 (br s, 1 H), 5.25–5.10 (pair of br d due to rotamers, 1 H, J = 7.5 Hz), 4.85-4.60 (pair of br m due

⁽¹⁷⁾ The analogous reduction using triethylsilane/TFA gave 23, 24, and 25 in a ratio of 77/9/14.

⁽¹⁸⁾ For previous syntheses of solenopsin A, see: (a) Mundy, B. P.; Bjorklund, M. Tetrahedron Lett. 1985, 3899. (b) Wasserman, H. H.; Rusiecki, V. Tetrahedron Lett. 1988, 29, 4977. (c) Padwa, A.; Chinn, R. L.; Zhi, L. Tetrahedron Lett. 1989, 30, 1491. (d) Nagasaka, T.; Hagashi, H.; Kumakawa, M.; Sakamoto, M. Heterocycles 1989, 29, 2157 and references therein.

⁽¹⁹⁾ For previous syntheses of dihydropinidine, see: (a) Hill, R. K.; Yuri, T. Tetrahedron 1977, 33, 1569 and references therein. (b) Astier, A.; Plat, M. M. Tetrahedron Lett. 1978, 2051. (c) Bonin, M.; Romero, J. R.; Grierson, D. S.; Husson, H.-P. Tetrahedron Lett. 1982, 23, 3369. Guerrier, L.; Royer, J.; Grierson, D. S.; Husson, H.-P. J. Am. Chem. Soc. 1983, 105, 7754.

to rotamers, 1 H), 1.50 (s, 9 H), 1.25 (br s, 20 H), 0.88 (t, 3 H, J = 6.6 Hz); ¹³C NMR (75 MHz, CDCl₃) δ 152.6 and 151.8 (due to rotamers), 127.7 and 127.5 (due to rotamers), 127.5 and 126.7 (due to rotamers), 117.8 and 117.4 (due to rotamers), 106.6 and 106.2 (due to rotamers), 81.7, 54.1 and 53.2 (due to rotamers), 34.3 and 33.8 (due to rotamers), 31.9, 29.69, 29.62, 29.57, 29.54, 29.50, 29.32, 28.1, 24.3, 22.7, 14.1; FT-IR (neat) 2955, 2926, 2854, 1718, 1633, 1369, 1388, 1171, 1127 1052 cm⁻¹.

1-(tert-Butoxycarbonyl)-4-chloro-6-methyl-2-n-undecyl-1,2-dihydropyridine (6). To a stirred solution of dihydropyridine 5 (3.20 g, 8.64 mmol) in 120 mL of THF at -42 °C was added 3.91 mL (10.33 mmol, 2.65 M solution in hexane) of n-butyllithium dropwise via syringe. After the mixture had stirred at -42 °C for 1 h, iodomethane (1.61 mL, 25.92 mmol) was added and stirring was continued at -42 °C for 1 h and then at room temperature for 1 h. Water (30 mL) and ether (60 mL) were added, the layers were separated, the aqueous phase was extracted with ether (2 \times 10 mL), and the combined organic extracts were washed with brine. The organic phase was dried over K₂CO₃, filtered through silica gel/Celite, and concentrated in vacuo to give the crude product. Purification by radial PLC (silica gel, 5% EtOAc/ hexanes) gave 2.74 g (83%) of 6 as a clear, light orange oil: ^{1}H NMR (300 MHz, $CDCl_3$) δ 5.62 (dd, 1 H, J = 6.6 Hz and J = 1.5Hz), 5.32 (m, 1 H, J = 1.5 Hz), 4.76 (dt, 1 H, J = 7.3 Hz and J= 6.6 Hz), 2.14 (s, 3 H), 1.50 (s, 9 H), 1.48–1.40 (m, 2 H), 1.26 (br s, 18 H), 0.88 (t, 3 H, J = 6.6 Hz); ¹³C NMR (75 MHz, CDCl₃) δ 153.1, 137.0, 126.6, 119.4, 112.3, 81.4, 54.2, 31.9, 31.8, 29.69, 29.63, 29.55, 29.44, 29.36, 29.33, 28.2, 24.6, 22.7, 22.0, 14.1; FT-IR (neat) 2954, 2923, 2852, 1709, 1637, 1471, 1393, 1368, 1342, 1169, 1131 cm⁻¹. Anal. Calcd for $C_{22}H_{38}CINO_2$: C, 68.81; H, 9.98; N, 3.65. Found: C, 69.09; H, 10.16; N, 3.27.

1-(tert-Butoxycarbonyl)-4-chloro-cis-6-methyl-2-n-undecyl-1,2,5,6-tetrahydropyridine (7). To a stirred solution of dihydropyridine 6 (100.6 mg, 0.262 mmol) in CH_2Cl_2 (10 mL) at -42 °C was added dropwise 0.25 mL (1.57 mmol) of triethylsilane followed by 0.202 mL (2.62 mmol) of trifluoroacetic acid. After stirring for 4 h at -42 °C, the cold bath was removed, and the reaction was immediately quenched with 10 mL of a saturated aqueous $NaHCO_3/THF$ mixture (50:50). The aqueous phase was extracted with CH_2Cl_2 (2 × 10 mL), and the combined organic extracts were washed with water (10 mL) and brine. The organic phase was dried over K₂CO₃, filtered through Celite, and concentrated in vacuo to give the crude product as a mixture of 75% cis isomer 7, 15% trans isomer 10, and 10% cis isomer 9 as determined by capillary GC and GC/MS analysis. Purification by radial PLC (silica gel, 5% EtOAc/hexanes) gave 79.5 mg (79%) of a colorless oil consisting of the three isomers in approximately the same ratio as found in the crude product. Isomer 7 had the following spectral data: ¹H NMR (300 MHz, CDCl₃) & 5.87 (m, 1 H), 4.67 (m, 1 H), 4.28 (m, 1 H), 2.80–2.65 (dm, 1 H, J = 16.9Hz), 2.05 (d, 1 H, J = 16.9 Hz), 1.60–1.45 (m, 2 H), 1.47 (s, 9 H), 1.26 (br s, 18 H), 1.21 (d, 3 H, J = 6.6 Hz), 0.88 (t, 3 H, J = 6.8Hz); ¹³C NMR (75 MHz, CDCl₃) δ 154.2, 127.5, 123.2, 79.7, 53.0, 44.4, 37.9, 36.3, 31.9, 29.6, 29.5, 29.3, 28.4, 26.9, 22.7, 21.0, 14.1; FT-IR (neat) 2927, 2855, 1697, 1458, 1406, 1392, 1366, 1343, 1177, 1104, 1070 cm⁻¹. Anal. Calcd for $C_{22}H_{40}ClNO_2$: C, 68.45; H, 10.44; N, 3.63. Found: C, 68.23; H, 10.22; N, 3.58.

1-(tert-Butoxycarbonyl)-4-chloro-trans-2-methyl-6-nundecyl-1,2,5,6-tetrahydropyridine (10). To a stirred solution of dihydropyridine 6 (402 mg, 1.047 mmol) in CH₂Cl₂ (50 mL) was added 395 mg (6.282 mmol) of sodium cyanoborohydride. After being stirred for 15 min at room temperature, the heterogeneous solution was cooled to -42 °C, and trifluoroacetic acid (0.81 mL, 10.47 mmol) was added dropwise. After stirring for 4 h at -42 °C, the cold bath was removed, and the reaction mixture was immediately quenched with 40 mL of a saturated aqueous NaHCO₃/THF mixture (50:50). The aqueous phase was exatracted with CH_2Cl_2 (2 × 15 mL), and the combined organic extracts were washed with water (20 mL) and brine. The organic phase was dried over K₂CO₃, filtered through Celite, and concentrated in vacuo to give the crude product as a mixture of 77% trans isomer 10, 12% cis isomer 7, 7% cis isomer 9, and 4% trans isomer 8 as determined by capillary GC and GC/MS analysis. Purification by MPLC (silica gel, 2% EtOAc/hexanes) gave 170.5 mg (combined mass) of isomers 10, 9, 8 (not completely separated), and 53 mg of 7 (223.5 mg total mass, 55%). Isomer 10 had the

following spectral data: ¹H NMR (300 MHz, CDCl₃) δ 5.81 (dd, 1 H, J = 5.1 Hz and J = 2.2 Hz), 4.24 (m, 1 H), 4.01 (br m, 1 H), 2.75–2.60 (dm, 1 H, J = 16.9 Hz), 2.29 (dd, 1 H, J = 16.9 Hz and J = 2.9 Hz), 1.60–1.45 (m, 2 H), 1.47 (s, 9 H) 1.31 (d, 3 H, J = 6.6 Hz), 1.26 (br s, 18 H), 0.88 (t, 3 H, J = 6.6 Hz); ¹³C NMR (75 MHz, CDCl₃) δ 155.2, 128.0, 127.1, 79.7, 52.9, 49.4, 35.5, 33.2, 31.9, 29.6, 29.4, 29.3, 28.4, 27.0, 22.6, 21.3, 14.1; IR (neat) 2920, 2845, 1695, 1460, 1365, 1315, 1250, 1165, 1125, 1085, 1050 cm⁻¹; HRMS calcd for C₂₂H₄₀ClNO₂ 385.2748, found 385.2746.

1-(tert-Butoxycarbonyl)-cis-2-methyl-6-n-undecylpiperidine (15) from 1,2,5,6-Tetrahydropyridine 7. Into an oven-dried, argon-flushed, 500-mL Parr bottle was placed an absolute ethanol solution (20 mL) of 1,2,5,6-tetrahydropyridine 7 (119.1 mg, 0.309 mmol). To the solution was added 120 mg of lithium carbonate and 120 mg of 5% Pd/C. The mixture was placed on a Parr hydrogenation apparatus and shaken for 12 h under 40 psi of hydrogen gas. The mixture was filtered through Celite and concentrated on a rotary evaporator. The residue was dissolved in ether (20 mL), water (10 mL) was added, the aqueous phase was extracted with ether $(2 \times 5 \text{ mL})$, and the combined ether extracts were washed with 1 N NaOH $(2 \times 10 \text{ mL})$ and brine. The organic phase was dried over MgSO₄, filtered through silica gel/Celite, and concentrated in vacuo to give the crude product. Kugelrohr distillation (0.25 mmHg, 170-185 °C) gave 91 mg (83%) of 15 as a colorless oil.

Preparation of 15 from 1,2-Dihydropyridine 6. Into an oven-dried, argon-flushed, 100-mL round-bottom flask was placed an absolute ethanol solution (20 mL) of 1,2-dihydropyridine 6 (264.5 mg, 0.687 mmol). To the solution was added 250 mg of lithium carbonate and 250 mg of 5% Pd/C. The mixture was placed under a positive pressure of hydrogen gas from a balloon and stirred for 12 h at room temperature. The mixture was filtered through Celite and concentrated on a rotary evaporator. The residue was dissolved in ether (20 mL), water (10 mL) was added, the aqueous phase was extracted with ether $(2 \times 5 \text{ mL})$, and the combined ether extracts were washed with 1 N NaOH (2×10 mL) and brine. The organic phase was dried over MgSO4, filtered through Celite, and concentrated in vacuo to give the crude product as a mixture of 98% cis isomer 15 and 2% trans isomer 16 as determined by capillary GC analysis. Purification by MPLC (silica gel, 5% EtOAc/hexanes) gave 201 mg (83%) of 15 as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 4.36-4.22 (m, 1 H), 4.08-3.98 (m, 1 H), 1.70-1.30 (m, 8 H), 1.46 (s, 9 H), 1.26 (br s 18 H), 1.15 (d, 3 H, J = 6.9 Hz), 0.88 (t, 3 H, J = 6.8 Hz); ¹³C NMR (75 MHz, CDCl₃) δ 155.3, 78.7, 50.3, 45.5, 35.1, 31.9, 30.3, 29.70, 29.66, 29.62, 29.60, 29.56, 29.3, 28.5, 27.6, 27.5, 22.6, 20.3, 14.1; FT-IR (neat) 2927, 2855, 1690, 1467, 1403, 1365, 1179, 1100, 1081 cm⁻¹. Anal. Calcd for $C_{22}H_{43}NO_2$: C, 74.73; H, 12.26; N, 3.96. Found: C, 74.99, H, 12.14; N, 3.82.

(±)-epi-Solenopsin A. cis-2-Methyl-6-n-undecylpiperidine (17). To a stirred solution of 192.5 mg (0.544 mmol) of Bocpiperidine 15 in 20 mL of CH₂Cl₂ at 0 °C was added dropwise 5 mL (excess) of trifluoroacetic acid. The cooling bath was removed, and stirring was continued for 1 h at room temperature. After concentrating the resulting solution on a rotary evaporator, the remaining liquid was dissolved in ether (20 mL), water (10 mL) was added, and the aqueous phase was extracted with ether $(2 \times 5 \text{ mL})$. The combined organic extracts were washed with saturated aqueous NaHCO₃ (10 mL) and brine. The organic phase was dried over K₂CO₃, filtered through Celite, and concentrated in vacuo to give the crude product. Purification by MPLC (silica gel, $Et_2O/1\%$ *i*-PrNH₂) gave 111.5 mg (81%) of 17 as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 2.68–2.56 (m, 1 H), 2.53–2.42 (m, 1 H), 1.81-0.94 (m, 9 H), 1.25 (br s, 18 H), 1.06 (d, 3 H, J =6.3 Hz), 0.88 (t, 3 H, J = 6.8 Hz); ¹³C NMR (75 MHz, CDCl₃) δ 57.0, 52.4, 37.4, 34.4, 32.2, 31.8, 29.76, 29.61, 29.57, 29.54, 29.52, 29.3, 25.9, 24.8, 23.0, 22.6, 14.0; FT-IR (neat) 3273 (weak), 2955, 2925, 2854, 2799, 2711, 1466, 1441, 1377, 1322, 1129 cm⁻¹

1-(*tert*-Butoxycarbonyl)-6-methyl-2-*n*-undecyl-1,2,3,4tetrahydropyridine (18). To a stirred solution of 740 mg (1.93 mmol) of 1,2-dihydropyridine 6 in 40 mL of MeOH at 0 °C was added 143 mg (1.93 mmol) of lithium carbonate followed by 148 mg (20 wt %) of 5% Pd/C. The mixture was placed under a positive pressure of hydrogen gas from a balloon, and the reaction progress was monitored by removing aliquots (0.10 mL) with a syringe, concentrating the aliquot in vacuo, and examining its ¹H NMR spectrum. (Under these conditions the reaction was complete in approximately 60 min.) Upon completion, the mixture was filtered through Celite and concentrated on a rotary evaporator. The residue was dissolved in ether (20 mL), water (10 mL) was added, the aqueous phase was extracted with ether (2×5) mL), and the combined ether extracts were washed with 1 N NaOH $(2 \times 10 \text{ mL})$ and brine. The organic phase was dried over K_2CO_3 , filtered through Celite, and concentrated in vacuo to give the crude product. Purification by radial PLC (silica gel, 5% EtOAc/hexanes/1% TEA) gave 515 mg (76%) of 18 as a clear, light yellow oil: ¹H NMR (300 MHz, $CDCl_3$) δ 4.86 (br s, 1 H), 4.39-4.31 (m, 1 H), 2.02 (m, 3 H), 1.99 (m, 2 H), 1.70 (m, 2 H), 1.48 (s, 9 H), 1.25 (br s, 20 H), 0.88 (t, 3 H, J = 6.8 Hz); ¹³C NMR (75 MHz, CDCl₂) δ 153.9, 132.6, 111.0, 80.0, 51.9, 31.9, 29.65, 29.62, 29.59, 29.57, 29.52, 29.33, 28.4, 26.4, 25.8, 23.1, 22.7, 19.5, 14.1; ¹H NMR (300 MHz, C_6D_6) δ 4.73 (br s, 1 H), 4.62 (m, 1 H), 2.21 (s, 3 H), 1.90–1.60 (m, 4 H), 1.45 (s, 9 H), 1.30 (br s, 20 H), 0.92 (t, 3 H, J = 6.5 Hz); ¹³C NMR (75 MHz, C₆D₆) δ 153.7, 133.4, 110.3, 79.6, 52.1, 32.3, 30.1, 30.0, 29.8, 28.4, 26.9, 26.2, 23.7, 23.1, 19.8, 14.4; FT-IR (neat) 2927, 2855, 1697, 1660, 1456, 1367, 1351, 1170, 1127, 1096, 1073 cm⁻¹. Anal. Calcd for C₂₂H₄₁NO₂: C, 75.16; H, 11.76; N, 3.98. Found: C, 74.90; H, 11.88; N, 3.71.

1-(tert-Butoxycarbonyl)-trans-2-methyl-6-n-undecylpiperidine (16) from 1,2,5,6-Tetrahydropyridine 10. Into an oven-dried, argon-flushed, 50-mL round-bottomed flask was placed an absolute ethanol solution (5 mL) of 1,2,5,6-tetrahydropyridine 10 (41.5 mg, 0.188 mmol). To this solution was added 42 mg of lithium carbonate and 42 mg of 5% Pd/C. The mixture was placed under a positive pressure of hydrogen gas from a balloon and stirred for 12 h at room temperature. The mixture was filtered through Celite and concentrated on a rotary evaporator. The residue was dissolved in ether (10 mL), water (5 mL) was added, the aqueous phase was extracted with ether $(2 \times 5 \text{ mL})$, and the combined ether extracts were washed with 1 N NaOH $(2 \times 5 \text{ mL})$ and brine. The organic phase was dried over K₂CO₃, filtered through Celite, and concentrated in vacuo to give the crude product. Pipet purification (silica gel, 5% EtOAc/hexanes) gave 30.1 mg (80%) of 16 as a colorless oil.

Preparation of 16 from 1,2,3,4-Tetrahydropyridine 18. To a stirred solution of 205 mg (0.584 mmol) of 1,2,3,4-tetrahydropyridine 18 in 40 mL of CH₂Cl₂ was added 220 mg (3.504 mmol) of sodium cyanoborohydride. After being stirred for 15 min at room temperature, the heterogeneous solution was cooled to -42 °C and trifluoroacetic acid (0.45 mL, 5.84 mmol) was added slowly dropwise. After stirring for 4 h at -42 °C, the cold bath was removed and the reaction mixture was immediately quenched with 30 mL of a saturated aqueous NaHCO₃/THF mixture (50:50). The aqueous phase was extracted with CH_2Cl_2 (2 × 10 mL), and the combined organic extracts were washed with water (10 mL) and brine. The organic phase was dried over K₂CO₃, filtered through Celite, and concentrated in vacuo to give 190 mg (92%) of the crude product as a mixture of 90% trans isomer 16 and 10% cis isomer 15 as determined by capillary GC analysis. This crude material was used directly in the next step (Purification by MPLC (silica gel, 5% EtOAc/hexanes) of the material from a separate reaction provided a sample for elemental analysis.): ¹H NMR (300 MHz, CDCl₃) δ 3.97–3.86 (m, 1 H), 3.83–3.72 (m, 1 H), 1.90-1.30 (m, 8 H), 1.46 (s, 9 H), 1.26 (br s, 18 H), 1.23 (d, 3 H, J = 6.6 Hz), 0.88 (t, 3 H, J = 6.8 Hz); ¹³C NMR (75 MHz, CDCl₃) & 155.2, 78.7, 51.6, 46.9, 34.3, 31.8, 29.57, 29.54, 29.51, 29.25, 28.5, 27.1, 26.8, 23.1, 22.6, 20.7, 14.0, 13.6; FT-IR (neat) 2926, 2855, 1690, 1467, 1393, 1366, 1178, 1090 cm⁻¹. Anal. Calcd for C22H43NO2: C, 74.73; H, 12.26; N, 3.96. Found: C, 74.65; H, 12.16; N, 3.96.

(±)-Solenopsin A. trans-2-Methyl-6-n-undecylpiperidine (1). To a stirred solution of 190 mg (0.537 mmol) of crude piperidine 16 (containing approximately 10% of cis isomer 15) in 20 mL of CH_2Cl_2 at 0 °C was added dropwise 5 mL (excess) of trifluoroacetic acid. The cooling bath was removed, and stirring was continued for 1 h at room temperature. After concentrating the resulting solution on a rotary evaporator, the remaining liquid was dissolved in ether (20 mL) and 10 mL of water was added. The aqueous phase was extracted with ether (2 × 5 mL), and the combined organic extracts were washed with saturated aqueous NaHCO₃ (2 × 15 mL) and brine. The organic phase was dried over K₂CO₃, filtered through Celite, and concentrated in vacuo to give the crude product as a mixture of 90% (±)-solenopsin A (1) and 10% (±)-epi-solenopsin A (17) as determined by capillary GC analysis. Separation and purification by MPLC (silica gel, Et₂O/1% *i*-PrNH₂) gave 94 mg (64% from compound 18) of pure 1 as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 3.12–3.00 (m, 1 H), 2.92–2.82 (m, 1 H), 1.70–1.15 (m, 9 H), 1.26 (br s, 18 H), 1.07 (d, 3 H, J = 6.3 Hz), 0.88 (t, 3 H, J = 6.8 Hz); ¹³C NMR (75 MHz, CDCl₃) δ 50.8, 45.8, 34.0, 32.9, 31.9, 30.7, 29.7, 29.59, 29.57, 29.3, 26.4, 22.6, 21.2, 19.5, 14.0; FT-IR (neat) 3292 (weak), 2931, 2853, 1465, 1376, 1141, 1067 cm⁻¹; (±)-solenopsin A·HCl mp 121–122 °C (CH₂Cl₂=Et₂O) (lit.^{18d} mp 114 °C). (±)-Solenopsin A·HCl mp 1.4-Cl. Anal. Calcd for C₁₇H₃₆ClN: C, 70.43; H, 12.52; N, 4.83. Found: C, 70.21; H, 12.39; N, 4.78.

4-Chloro-1-(phenoxycarbonyl)-2-n-propyl-1,2-dihydropyridine (20). To a stirred solution of 4-chloropyridine (1.17 g, 10.30 mmol) in 75 mL of THF at -78 °C was added npropylmagnesium chloride (5.15 mL, 10.30 mmol, 2.0 M solution in diethyl ether) followed, after 20 min, by the dropwise addition of phenyl chloroformate (1.29 mL, 10.30 mmol). Stirring was continued for 30 min at -78 °C; then the cooling bath was removed and the reaction mixture was allowed to stir for 30 min longer while being slowly warmed to room temperature. Aqueous 20% NH₄Cl (30 mL) and ether (40 mL) were added, the layers were separated, and the aqueous phase was extracted with ether $(2 \times$ 25 mL). The combined organic extracts were washed successively with 25-mL portions of saturated aqueous CuSO4, water, saturated aqueous NaHCO₃, and brine. The organic phase was dried over MgSO₄, filtered through Celite, and concentrated in vacuo to give 3.08 g (quantitative) of crude product 20 as a clear, light yellow oil (This crude material was used directly in the next step to make the N-Boc derivative.): ¹H NMR (300 MHz, CDCl₃) § 7.45-7.10 (m, 5 H), 6.95-6.85 (pair of d due to rotamers, 1 H, J = 8.1 Hz), 5.68 (dd, 1 H, J = 6.2 Hz and J = 2.2 Hz), 5.40–5.28 (pair of dd due to rotamers, 1 H, J = 8.1 Hz and J = 2.2 Hz), 5.05–4.85 (pair of m due to rotamers, 1 H, J = 6.2 Hz), 1.85-1.30 (m, 4 H), 0.94(m, 3 H); FT-IR (neat) 3102, 3072, 3045, 2960, 2933, 2873, 1730, 1636, 1593, 1495, 1420, 1358, 1333, 1202, 1043 cm⁻¹.

1-(tert-Butoxycarbonyl)-4-chloro-2-n-propyl-1,2-dihydropyridine (21). By use of a procedure similar with that described for the preparation of 5, 3.08 g (10.30 mmol) of 20 was treated with 5.0 g (44.0 mmol) of potassium tert-butoxide. Purification by radial PLC (silica gel, 5% EtOAc/hexanes) gave 2.47 g (93%) of 21 as a colorless oil: ¹H NMR (300 MHz, $\overline{CDCl_3}$) δ 6.90-6.60 (pair of br d due to rotamers, 1 H, J = 7.3 Hz), 5.55 (br s, 1 H), 5.25-5.05 (pair of br d due to rotamers, 1 H, J = 7.3Hz), 4.90-4.60 (pair of br m due to rotamers, 1 H), 1.70-1.25 (m, 4 H), 1.50 (s, 9 H), 0.91 (t, 3 H, J = 7.3 Hz); ¹³C NMR (75 MHz, CDCl₃) & 152.6 and 151.7 (due to rotamers), 127.7 and 127.5 (due to rotamers), 127.4 and 126.7 (due to rotamers), 117.7 and 117.3 (due to rotamers), 106.6 and 106.1 (due to rotamers), 81.7, 53.7 and 52.9 (due to rotamerrs), 36.3 35.8 (due to rotamers), 28.1, 17.4, 13.9; FT-IR (neat) 2961, 2934, 2874, 1715, 1633, 1370, 1338, 1311, 1253, 1172, 1127, 1051 cm^{-1} .

1-(*tert*-Butoxycarbonyl)-4-chloro-6-methyl-2-*n*-propyl-1,2-dihydropyridine (22) was prepared by a procedure similar to that described for the preparation of 6. To 2.06 g (8.0 mmol) of 21 was added 3.95 mL (9.60 mmol, 2.43 M solution in hexane) of *n*-butyllithium followed after 1 h by 1.50 mL (24.0 mmol) of iodomethane. Purification by radial PLC (silica gel, 10% Et-OAc/hexanes) gave 1.83 g (84%) of 22 as a clear, light yellow oil: ¹H NMR (300 MHz, CDCl₃) δ 5.62 (dd, 1 H, J = 6.6 Hz and J= 1.5 Hz), 5.33 (m, 1 H, J = 1.5 Hz), 4.80 (dt, 1 H, J = 7.3 Hz and J = 6.6 Hz), 2.14 (s, 3 H), 1.52-1.24 (m, 4 H), 1.50 (s, 9 H), 0.91 (t, 3 H, J = 7.3 Hz); ¹³C NMR (75 MHz, CDCl₃) δ 153.0, 137.0, 126.6, 119.3, 112.2, 81.3, 53.9, 33.9, 28.1, 21.9, 17.8, 13.9; FT-IR (neat) 2962, 2932, 2874, 1708, 1637, 1393, 1369, 1343, 1328, 1173, 1132, 1086 cm⁻¹. Anal. Calcd for C14H₂₂ClNO₂: C, 61.87; H, 8.16; N, 5.15. Found: C, 61.66; H, 8.16; N, 5.26.

1-(tert-Butoxycarbonyl)-4-chloro-cis-6-methyl-2-npropyl-1,2,5,6-tetrahydropyridine (23). By use of a procedure similar with that described for the preparation of 7, 1.0 g (3.68 mmol) of 22 was treated with 4.61 mL (22.08 mmol) of tri-*n*propylsilane and 2.84 mL (36.80 mmol) of trifluoroacetic acid. The crude product consisted of a mixture of 80% of cis isomer 23, 12% of trans isomer 25, and 8% of cis isomer 24 as determined by capillary GC analysis. Purification by MPLC (silica gel, 5% EtOAc/hexanes) gave 722 mg (72%) of pure 23 as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 5.87 (dd, 1 H, J = 3.75 Hz and J = 2.85 Hz), 4.68 (m, 1 H), 4.31 (m, 1 H), 2.72 (dm, 1 H, J = 16.8Hz), 2.06 (d, 1 H, J = 16.8 Hz), 1.70–1.30 (m, 4 H), 1.47 (s, 9 H), 1.21 (d, 3 H, J = 7.2 Hz), 0.94 (t, 3 H, J = 7.2 Hz); ¹³C NMR (75 MHz, CDCl₃) δ 154.2, 127.5, 123.1, 79.8, 52.7, 44.5, 38.4, 37.9, 28.4, 20.9, 20.0, 13.9; FT-IR (neat) 2966, 2934, 2874, 1695, 1479, 1457, 1407, 1367, 1343, 1318, 1177, 1116, 1100, 1070 cm⁻¹. Anal. Calcd for C₁₄H₂₄ClNO₂: C, 61.41; H, 8.84; N, 5.12. Found: C, 61.50; H, 9.07; N, 5.17.

1-(tert-Butoxycarbonyl)-cis-2-methyl-6-n-propylpiperidine (26) from 1,2,5,6-Tetrahydropyridine 23. By use of a procedure similar with that described for the preparation of 15, a mixture of 587 mg of tetrahydropyridine 23 (2.14 mmol), 590 mg of lithium carbonate, and 590 mg of 10% Pd/C in 30 mL of ethyl acetate was hydrogenated for 2 days. Purification of the crude product by MPLC (silica gel, 10% EtOAc/hexanes) gave 428 mg (83%) of 26 as a colorless oil.

Preparation of 26 from 1,2-Dihydropyridine 22. By use of a procedure similar with that described for the preparation of 15, a mixture of 304 mg of dihydropyridine 22 (1.12 mmol), 300 mg of lithium carbonate, and 300 mg of 5% Pd/C in 20 mL of ethyl acetate was hydrogenated for $\bar{2}$ days. (Note: Shorter reaction times gave the incomplete reduction product 1,2,3,4tetrahydropyridine as a significant contaminant.) The crude product consisted of a mixture of 98% cis isomer 26 and 2% of the trans isomer as determined by capillary GC analysis. Purification by MPLC (silica gel, 10% EtOAc/hexanes) gave 187 mg (69%) of 26 as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 4.36-4.22 (m, 1 H), 4.10-4.00 (m, 1 H), 1.70-1.20 (m, 10 H), 1.46 (s, 9 H), 1.16 (d, 3 H, J = 7.2 Hz), 0.92 (t, 3 H, J = 7.3 Hz); ¹³C NMR (75 MHz, CDCl₃) δ 155.2, 78.7, 50.0, 45.5, 37.2, 30.2, 28.4, 27.3, 20.6, 20.3, 14.1, 14.0; FT-IR (neat) 2960, 2936, 2872, 1690, 1458, 1404, 1391, 1365, 1348, 1180, 1103, 1083 cm⁻¹. Anal. Calcd for C₁₄H₂₇NO₂: C, 69.67; H, 11.27; N, 5.80. Found: C, 69.64; H, 11.27; N, 5.80.

(±)-Dihydropinidine. cis-2-Methyl-6-n-propylpiperidine (2). To a stirred solution of 226 mg (0.936 mmol) of Boc-piperidine 26 in 15 mL of acetonitrile was added sodium iodide (562 mg, 3.744 mmol) followed by chlorotrimethylsilane (0.48 mL, 3.744 mmol). After stirring for 12 h at room temperature, saturated aqueous K_2CO_3 (5 mL) was added and stirring was continued for 1 h. The layers were separated, the aqueous phase was extracted with ether $(3 \times 5 \text{ mL})$, and the combined organic extracts were dried over K₂CO₃. Filtration through Celite and concentration on a rotary evaporator with the water bath at 0 °C gave the crude product. Kugelrohr distillation (60-100 °C, 20 mmHg (water aspirator), receiver bulb at -78 °C) gave 93 mg (70%) of 2 as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 2.68-2.56 (m, 1 H), 2.55-2.45 (m, 1 H), 1.76 (m, 1 H), 1.60 (m, 2 H), 1.55–0.80 (m, 8 H), 1.06 (d, 3 H, J = 6.6 Hz), 0.91 (m, 3 H); ¹³C NMR (75 MHz, CDCl₃) δ 56.8, 52.5, 39.7, 34.4, 32.2, 24.9, 23.1, 19.1, 14.3; ¹H NMR (300 MHz, $C_6D_6)$ δ 2.52–2.42 (m, 1 H), 2.40–2.32 (m, 1 H), 1.69 (m, 1 H), 1.48 (m, 2 H), 1.35-0.80 (m, 8 H), 0.99 (d, 3 H, J = 6.3 Hz), 0.88 (t, 3 H, J = 6.3 Hz); ¹³C NMR (75 MHz, C₆D₆) δ 57.1, 52.7, 40.1, 34.8, 32.5, 25.4, 23.2, 19.3, 14.6; FT-IR (neat) 3276 (weak), 2957, 2928, 2857, 2798, 2713, 1463, 1441, 1377, 1321, 1129 cm⁻¹; (±)-di-hydropinidine-HCl mp 212–213 °C (2:1 EtOAc–EtOH) (lit. 19e mp 210-213 °C).

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Registry No. (±)-1, 28720-60-7; (±)-1·HCl, 63950-17-4; (±)-2, 65337-42-0; (±)-2·HCl, 121963-72-2; 3, 626-61-9; (±)-4, 132410-08-3; (\pm) -5, 132410-09-4; (\pm) -6, 132438-34-7; (\pm) -7, 132410-10-7; (\pm) -8, 132410-11-8; (±)-9, 132410-12-9; (±)-10, 132410-13-0; (±)-15, 132410-14-1; (\pm) -16, 132410-15-2; (\pm) -17, 63950-16-3; (\pm) -18, 132410-16-3; (±)-20, 132410-17-4; (±)-21, 132410-18-5; (±)-22, 132410-19-6; (±)-23, 132410-20-9; (±)-24, 132410-21-0; (±)-25, 132410-22-1; (\pm) -26, 132410-23-2; Br(CH₂)₁₀CH₃, 693-67-4.

Thermal and Lewis Acid Induced Cycloaddition of Thioaldehyde S-Oxides (Monosubstituted Sulfines) to Dienes. 3^{1,2}

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The stereochemical course of the 1,4-cycloaddition of thioaldehyde S-oxides (monosubstituted sulfines) to 2,3-dimethylbuta-1,3-diene, buta-1,3-diene, and cis- and trans-penta-1,3-diene was investigated. Unexpectedly, the reactions of buta-1,3-diene and 2,3-dimethylbuta-1,3-diene with Z-monoaryl sulfines afforded cis/trans mixtures of the corresponding dihydrothiopyran S-oxides, in which the relative amounts of the two isomers depended upon the initial diene/sulfine ratio. A Z to E isomerization of the dienophiles during the cycloaddition was responsible. On the other hand, Z/E mixtures of aliphatic tert-butyl sulfine gave, with 2,3-dimethylbuta-1,3-diene, only the corresponding trans cycloadduct. Catalysis of the reaction by Lewis acids, heretofore largely unexplored, was also investigated.

Introduction

Considerable attention has been payed to the chemistry of disubstituted sulfines, the literature of which has been recently reviewed.³ In particular, the stereochemical

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course of the cycloaddition of unsymmetrically disubstituted sulfines to dienes has been investigated in detail,

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 Maccagnani, G.; Zani, P. J. Org. Chem. 1990, 55, 3744.
 Bonini, B. F.; Mazzanti, G.; Zani, P.; Maccagnani, G.; Barbaro, G.;
 Battaglia, A.; Giorgianni, P. J. Chem. Soc., Chem. Commun. 1986, 964.